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Key indicators

Single-crystal X-ray study T = 173 K Mean σ (C–C) = 0.002 Å R factor = 0.035 wR factor = 0.096 Data-to-parameter ratio = 14.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2-(5-Ethyl-3-methylsulfanyl-1-benzofuran-2-yl)-acetic acid

The title compound, $C_{13}H_{14}O_3S$, was prepared by alkaline hydrolysis of ethyl 2-(5-ethyl-3-methylsulfanyl-1-benzofuran-2-yl)acetate. The crystal structure is stabilized by aromatic π - π stacking interactions and inversion-related intermolecular $O-H\cdots O$ hydrogen bonds between adjacent carboxyl groups.

Comment

This work is related to our previous communications on (3methylsulfanyl-1-benzofuran-5-yl)acetic acid derivatives (Choi *et al.*, 2006, 2007), with the purpose of understanding how the molecular conformation and crystal structure are affected by acetic acid substituents on the 1-benzofuran ring system. The title compound, (I) (Fig. 1), was obtained by hydrolysis of ethyl 2-(5-ethyl-3-methylsulfanyl-1-benzofuran-2-yl)acetate with potassium hydroxide.



In (I), bond lengths and angles are as expected for this type of compound (Choi *et al.*, 2006, 2007). In the crystal structure (Fig. 2), $\pi - \pi$ stacking interactions are observed between the furan and benzene rings of adjacent benzofuran systems, with $Cg1 \cdots Cg2^{i} = 3.801$ (2) Å (Cg1 and Cg2 are the centroids of the O1/C8/C1/C2/C7 and C2–C7 rings; symmetry code as in Fig. 2). Classical inversion-related O2—H14···O3ⁱⁱⁱ hydrogen bonds link the carboxyl groups of adjacent molecules (Table 1 and Fig. 2).

Experimental

Ethyl 2-(5-ethyl-3-methylsulfanyl-1-benzofuran-2-yl)acetate (834 mg, 3 mmol) was added to a solution of potassium hydroxide (561 mg, 10.0 mmol) in water (10 ml) and methanol (10 ml); the mixture was heated at 333 K for 4 h and then cooled. Water (20 ml) was added, and the solution was washed with dichloromethane. The aqueous layer was acidified to pH = 1.0 with concentrated hydrochloric acid and then extracted with chloroform, dried over magnesium sulfate, filtered and concentrated under vacuum. The residue was purified by column chromatography (hexane–ethyl acetate, 1:2 ν/ν) to give (I) as a white solid. Crystals suitable for X-ray analysis were grown by slow evaporation of a benzene solution [yield 92%, m.p. 396–397 K; $R_{\rm F} = 0.46$ (hexane–ethyl acetate, 1:2 ν/ν)].

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Crystal data

 $C_{13}H_{14}O_3S$ $M_r = 250.30$ Triclinic, $P\overline{1}$ a = 5.0746 (4) Å b = 8.0636 (6) Å c = 15.058 (1) Å $\alpha = 94.711(1)^{\circ}$ $\beta = 93.153 (1)^{\circ}$

Data collection

Bruker SMART CCD diffractometer Absorption correction: none 3461 measured reflections

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.035$ H atoms treated by a mixture of $wR(F^2) = 0.096$ S = 1.10 $\Delta \rho_{\text{max}} = 0.28 \text{ e} \text{ Å}^{-3}$ 2368 reflections $\Delta \rho_{\rm min} = -0.22 \text{ e} \text{ Å}^{-3}$ 159 parameters

Table 1

Hydrogen-bond geometry (Å, °).

$D-\mathrm{H}\cdots A$	<i>D</i> -H	Н…А	$D \cdots A$	$D - H \cdot \cdot \cdot A$
$O2-H14\cdots O3^i$	0.82 (3)	1.87 (3)	2.684 (2)	175 (3)

 $\gamma = 90.538 \ (1)^{\circ}$

Z = 2

V = 613.10 (8) Å³

Mo $K\alpha$ radiation

 $0.32 \times 0.25 \times 0.20$ mm

2368 independent reflections

2145 reflections with $I > 2\sigma(I)$

independent and constrained

 $\mu = 0.26 \text{ mm}^-$

T = 173 (2) K

 $R_{int} = 0.031$

refinement

Symmetry code: (i) -x + 2, -y + 2, -z + 1.

Atom H14 of the carboxyl group was found in a difference Fourier map and refined freely with an isotropic displacement parameter. The other H atoms were positioned geometrically and refined using a riding model, with C-H = 0.95 Å for aromatic H atoms, 0.98 Å for methyl H atoms and 0.99 Å for methylene H atoms, and with $U_{iso}(H) = 1.2U_{ea}(C)$ for aromatic and methylene H atoms and $1.5U_{eq}(C)$ for methyl H atoms.

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997) and DIAMOND (Brandenburg, 1998); software used to prepare material for publication: SHELXL97.

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Figure 1

The molecular structure of (I), with displacement ellipsoids drawn at the 50% probability level for non-H atoms.



Figure 2

Hydrogen-bond and $\pi - \pi$ interactions (dotted lines) in (I). [Symmetry codes: (i) 1 + x, y, z; (ii) x - 1, y, z; (iii) 2 - x, 2 - y, 1 - z.]

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